

Atomic Mass Energy and Constant

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Abstract:

We know that when different methods are used and same result happens, then, that is a 'proof' in science language.

In this Study, When, I decided to calculate different units (Calorie, Joule, eV, etc) then, I found difficulty in calculator's overview, error etc. features,

Therefore, I thought if 1kg of pure water (4°C,) = one liter Water = 1 Kg ice and 1 liter water = cubic decimeter. etc.

Then why I could not measure 1.782662×10^{-36} Kg. Mass = $1.6021766208 \times 10^{-19}$ Joule energy. In that place 1.782662 unit mass = 1.6021766208 unit energy.

After That all differential equation is obtained under the assumption fractional formula. In this process, I have shown all calculations in my present and previous essays. If anyone found difficulty therefore, I have shown present units (Calorie, Joule, eV, etc) how to be compared with example. Previously, the digits or, numbers are calculated are by no means have a relation with the present units and I found same result is coming with different methods then, I have given impetus on the number to get the next result.

Introduction:

In this article and previous article **WILL ORIGINATE A TECHNIQUE** which gives a new process of calculation about atomic structure.

To discuss on this large hypothesis (Subject), this understanding is felt that to get full success or, to complete it is a very difficult task. To maintain quality and a unilateral position by including everyone's opinion is really an arduous job.

. To do this work, here the help of image mode is taken due to typing error. In present essay, have shown the digit or number's used with its source and how it is Inter-related with each ether. The quantum digits are not abstract or, mind creation, I have tried to make the result by scientific knowledge. For example quantum digit of Red – 3, Yellow – 4, etc. the numbers are used for different lights for calculation of light wave. The source's are discussed like wise Mass, (like 1.67..., 1.679...; 1.688...; average Value of mass and Mass, like 1.777...; 1.778...; 1.78...; Details discussed by images in this article); Energy (like E, 1.50...; 1.501..., e = 1.602...; 1.604...; Magnetic Field 1.44...; Unit. Etc.) different numbers of speed's are put together also have been shown.

By comparing the result with present unit, by calculating here with comparing present unit which will be easily accessible to all kind of readers and I have given special attention as it.

Here, it is found that, by comparing with 'Planck's Constant from different Mass to Energy, or, Energy to Mass, not only Honorable Scientist 'Planck's Constant, **DIFFERENT type of SIX CONSTANT** are present, and we can also create new rule from the relation.

[Here also, it is shown that in present and previous article the numbers or digit/units are used, how they have come, and in the process of getting those unit and unit without power (calculated) is almost same. If we need to understand completely then, we have to read consciously every essay. ('Atomic Mass Energy Primary Colour', 'Atomic Time Space', 'Atom To Nucleus', Published in most reliable media Mendeley Data, Datacite; Narcis.nl. & A Story Of Latest Scientific Discovery. All articles also Published in great social media like Blog, Twitter, Facebook,) This total calculation should be multiplied (or divided) by 10 or, multiplier of 10 like 1000,10000,1,00,000 to convert it from current unit to different unit like- Kilogram, gram, meter, second, etc. As per need]

1.1- Mathematical Equation

Shown present Unit Calorie, Joule, eV, etc. How to be compared with proved digits.

$\sqrt[6]{9}$ or $\sqrt[3]{3} = 1.44224957$ (Joule) = $9 (\times 10^{+18} \text{ eV})$; $1.44224957 \times (10/9) = 1.602499$ Unit (Energy),
 $1.602499 \times (10/9) = 1.7805555$ Unit (Mass Before create a Nucleus),

$1.44224957 \text{ eV} = 2.31 (\times 10^{-19}) \text{ Joule}$. $2.31^2 = 5.3361$ Unit (Force),
 $\sqrt{2.31} = 1.5198684\dots \times (10/9) = 1.68874268373007373685208268956, \text{unit}$.

$[5.3361 / 3 = 1.7787, 1.7787^2 = 3.16377369 \text{ Unit (for acceleration)}; 3.16377369^2 = 10; < - > 5.3361 \times 3 = 16.0083; 16.0083 / 9 = 1.7787, \sqrt{1.7787} = 1.333679 = (4/3); 5.3361 \times (10) = 53.361; 53.361 / 9 = 5.929,$
 $\sqrt[4]{5.929} = 1.56043384939\dots \times (10/9) = 1.73381; 1.73381^2 = 3,$
 $\sqrt[4]{5.3361} = 1.519868415357\dots \times (10/9) = 1.68874268373.. \text{Unit mass (After create a Nucleus),}$
 $\sqrt{1.68874} = 1.299,$ will be discuss in following chapter]

$1.78 \text{ eV} = 2.8518742234 \times 10^{-19} \text{ Joule}$, $\sqrt{2.8518742234} = 1.68874, \text{Unit (After create a Nucleus)}$, $1.78 \text{ Joule} = 1.110988687370174 \times 10^{+19} \text{ eV}$,
 $1.110988687370174 = (10/9)$, $\sqrt{1.11098868} = 1.05403448.. = (M / m.)^{1/2}$, (Please see Past article)

$\sqrt[3]{3} = 1.44224957\dots; 1.44224957 \times (0.9) = \mathbf{1.298 \text{ (Constant)}};$
 [will be discussed in following chapters]

$\sqrt{3} \times (3/4) = \mathbf{1.299 \text{ (Constant)}}; (\sqrt[6]{3}) / (3/4) = \mathbf{1.60124927356\dots; \text{ (energy)}}$

$\sqrt[3]{3} = 1.44224957\dots / (7/22) = 4.53278436382\dots;$

$\sqrt[3]{4.53278436382} = 1.654963245086\dots; \sqrt{1.654963245086\dots} = \mathbf{1.2864537};$

$1.2864537\dots \times 7 = \mathbf{9}$; $1.44224957\dots \times 9 = 12.98024$; $\sqrt[3]{12.98024} = 2.3504\dots;$

$\sqrt{2.3504} = 1.533017\dots; 1.533017 \times (10/9) = 1.70335294285\dots; 1.70335294285 \times 2 = 3.4067058857,$

$\sqrt[3]{3.4067058857} = 1.5046825356149681559192516970299 \times (10/9) = 1.6718694840166 \text{ (m)}$

$(7/22) / (3/4) = 0.4242424\dots; (r^{-3}); \text{ as per } (4/3) \times \text{Pi} \times r^3; = 4.18879 \times 0.424242 = 1.777 \text{ (M)}, 0.424242\dots \times 4 = 1.69696\dots \text{ (m)}$;

$(3/4) / (7/22) = 2.3571428571\dots; \sqrt[3]{2.35714285\dots} = 1.3308 (=4/3), \text{sq} > 1.3308 = 1.77102864 \text{ (M)}$

$\sqrt{2.3571428571\dots} = 1.535298947157\dots;$

$1.535298947157\dots \times (10/9) = 1.70588771906\dots; 1.70588771906\dots \times 2 = 3.411775438\dots;$

$\sqrt[3]{3.411775438} = 1.5054285433..$; $\sqrt{1.505428..} = 1.239..$; (ev)
 $1.5054285433.. \times (10/9) = 1.67269838..$; (m); [we know one Neutron Mass = $1.6749.. \times 10^{-27}$ Kg. And Proton Mass = $1.6726219 \times 10^{-12}$ Kg.] [will be discussed in following chapters]

That means it is seen that Mass is present in two positions. One is Nucleus before birth & another nucleus after birth – Before the creation of nucleus the Mass is more and energy is less, after that Nucleus after creation Mass get lesser and energy gets bigger. The less amount of mass is converted into energy. So, here 'M' is the Mass before birth Nucleus, and 'm' is the mass of after birth Nucleus. And also it is seen that Force = 5.3361 Unit, Magnetic Field Either Wave = 1.44224957, Unit, 'f' or acceleration = 3.16377369 & Energy = 1.602499 Unit . Time for acceleration = 1.11098868 when 'v' = 3, present in a particle (Reference 'Atomic Mass Energy Primary Colour' 'A Story Of Latest Scientific Discovery')

1.2

..Hence ; This Calculation Done By Each Colors Frequency's Equivalent Mass to Calorie, eV, Joule, [Let us see by Identical Result; if we convert each Frequency Joule to Electron Volt then –(This calculation / Table Done in past article "A Story of Latest Scientific Discovery" & "Atomic Mass Energy Primary Colours" here is shown that '**L**' = **0.413334** involve in each calculation in all articles Please see.)]

TABLE : 1

Red fqn. = 3.8757 ,
 3.8757 Joule equivalent eV = $2.419021829011563 (x10^{+19})$ eV.
 $2.419021829011563 \times 0.413334 = \underline{1 \text{ unit}}$,
 Orange fqn. = 4.5216, 4.5216 Joule = $2.822160926299426 (x10^{+19})$ eV.
 $2.822160926299426 \times 0.413334 = \underline{1.1665 \text{ unit}}$
 Yellow fqn. = 5.1676, 5.1676 joule = $3.22536243868208 (x10^{+19})$ eV
 $3.22536243868208 \times 0.413334 = \underline{1.3331 \text{ unit}}$
 Green fqn. = 5.4905, 5.4905 joule = $3.426900779778618 (x10^{+19})$ eV
 $3.426900779778618 \times 0.413334 = \underline{1.4164 \text{ unit}}$
 Cyan fqn. = 5.8135 , 5.8135 joule = $3.628501535969947 (x10^{+19})$ eV
 $3.628501535969947 \times 0.413334 = \underline{1.4997 \text{ unit}}$
 Blue fqn. = 6.1365 ,6.1365 Joule = $3.830102292161276 (x10^{+19})$ eV
 $3.830102292161276 \times 0.413334 = \underline{1.5831 \text{ unit}}$

Total-- 7.9988 + 1 = 8.9988 or 9, [8.99...(i) in this Chapter]
(Red = 1 unit, one another Red present in T1 group See Past articles,)
[T 3 Group Violet fqn. = 7.7514, 7.7514 joule = $4.838043658023127 (x10^{+19})$ eV
 $4.838043658023127 \times 0.413334 = 1.9997$]

Concept of used Number / Digit as per Table : 1

Total: T1- (1+1.1665+1.3331 = 3.4996 Unit, T2- (1.4164+1.4997+1.5831) = 4.4992 Unit.

TABLE: (a)

T1 Group:

- *(1) 3.4996 Cal. = 9.14513788315199e+19 eV.
- *(2) 3.4996 J. = 2.184278657483517e+19 eV.
- *(3) 3.4996 eV. = 1.339203445205885e-19 cal.
- *(4) 3.4996 J. = 0.8358650998375848 cal.
- *(5) 3.4996 cal. = 14.65212528, J.
- *(6) 3.4996 eV. = 5.606976984388e-19 J.

T2 Group:

- *(7) 4.4992 Cal = 11.75728779399858e+19 eV.
- *(8) 4.4992 J. = 2.808179945065105e+19 eV.
- *(9) 4.4992 eV = 1.72172366575332e-19 cal,
- *(10) 4.4992 Joule = 1.07461545810643 cal.
- *(11) 4.4992 cal = 18.83725056 J.
- *(12) 4.4992 eV = 7.208512643776e-19 J.

[As per TABLE : (a), T1 & T2 Groups Total, eV, Joule, Calorie; & Discussion about Digit of Mass, energy, Magnetic field/Wave etc. & it's inter relation]

$$\begin{aligned}
 &T1 > (9.14513788315199e+19 + 2.184278657483517e+19 + 1.339203445205885e-19 + \\
 &0.8358650998375848 + 14.65212528 + 5.606976984388e-19) = \\
 &1.1329416540635507001548799037984e+20 \text{ Unit. --- } (T_1 \text{ eV} + \text{eV} + \text{Cal} + \text{Cal} + \text{J} + \text{J}) \\
 &\sqrt[9]{1.1329416540635507001548799037984e + 20} = 1.691395935863731528846735405114e+2 \\
 &\sqrt[81]{1.1329416540635507001548799037984e + 20} = 1.7684093774173502175912436551843(M1) \\
 &(1.1329416540635507001548799037984e + 20)^2 = 1.2835567915122541866966940802516e+40 \\
 &1.2835567915122541866966940802516e+40 \times 7 = 8.9848975405857793068768585617613e+40 (C^2)
 \end{aligned}$$

$$\begin{aligned}
 &\text{Total, } T2 > (11.75728779399858e+19 + 2.808179945065105e+19 + 1.72172366575332e-19 + \\
 &1.07461545810643 + 18.83725056 + 7.208512643776e-19) = \\
 &1.4565467739063685001991186601811e+20 \text{ Unit. --- } (T_2 \text{ eV} + \text{eV} + \text{Cal} + \text{Cal} + \text{J} + \text{J}) \\
 &\sqrt[81]{1.4565467739063685001991186601811e + 20} = 1.7739032577847272685285436494467(M_2) \\
 &(1.7739032577847272685285436494467)^2 = 3.1467327679792685646147912404126 (= \text{Pi}) \\
 &1.7739032577847272685285436494467 (M_2) + 1.7684093774173502175912436551843 (M_1) \\
 &= 3.542312635202077486119787304631, \\
 &(3.542312635202077486119787304631)^2 = 12.5479788055122865, = (4\text{Pi.})
 \end{aligned}$$

$$\begin{aligned}
 &[(T_2, \text{Total}) / (T_1, \text{Total})] = \\
 &(1.4565467739063685001991186601811e+20 / 1.1329416540635507001548799037984e+20) \\
 &= 1.2856326437307120544338840572299 \times 7 = 8.999428506115, (C^2 \text{ either Speed of light sqr.}) \\
 &[(T_1, \text{Total}) \times (T_2, \text{Total})] = \\
 &(1.4565467739063685001991186601811e+20 \times 1.1329416540635507001548799037984e+20) \\
 &= 1.6501825112504097370391701010739e+40, \\
 &\sqrt{1.6501825112504097370391701010739e + 40} = 1.2845942983099410195821633117097e+20 \\
 &1.2845942983099410195821633117097e+20 \times 7 = 8.9921600881695871370751431819682e+20 (C^2)
 \end{aligned}$$

From TABLE: (a), In T1 Groups have total 6 no. of 3.4996, unit.

And in T2 Groups have total 6 no. of 4.4992, unit.

So, 3.4996 x 6 = 20.9976,

$\sqrt[6]{20.9979} = 1.660969316545.$ unit (almost) Same digit of Atomic Mass.

4.4992 x 6 = 26.9952,

$\sqrt[6]{26.9952} = 1.731999483743053..; = \sqrt{3}; (\sqrt{26.9952} = 5.19569 = (\sqrt{3})^3)$

$$(3.4996)^2 + (4.4992)^2 = 32.4900008$$

$$3.4996 + 4.4992 = 7.9988, \sqrt[4]{7.9988} = 1.68172976.$$

$$7.9988 \times 2 = 15.9976 / 9 = 1.777511\dots;$$

$$4.4992 / 3.4996 = 1.28563264373\dots; \times 7 = 8.999\dots; \text{ or '9' [(ii) } C^2, \text{ or Speed of Light sq.]}$$

$$7.9988 \text{ eV} = 1.2815489628164\text{e-18 Joule.}$$

[About 1.28...; & 8.9...; Digit have explain with example/ Equation in present and past article.

This Type of example Will be Discussed in following Chapters which is shown by images]

2.1

As per Calculation of Table:1

By ensuring this calculation, Each Group's have three Part. Here Sections.

[We Know Planck's Constant, $6.62607004(80) \times 10^{-34}$ Js. $4.145667662(25) \times 10^{-15}$ eV.s]

T₁- Group

[A] When **Red** Mass 1 unit.

[Bar section = 1]

$$1 \text{ Calorie} = 2.61319518892216\text{e+19. eV}$$

$$1 \text{ Calorie} = 4.1868 \text{ Joule, } 4.1868 = 4/3 \text{ Pi. Or } 4.1868 / (4/3) = \text{Pi.}$$

[Frequency & Energy section = 2]

$$1 \text{ eV} = 3.826732898633801\text{e-20 .Calorie. } > \text{ This is RED Frequency.}$$

$$1 \text{ eV} = 1.60217653\text{e-19 Joule}$$

[Quantum section = 3]

$$1 \text{ Joule} = 0.2388458966274959. \text{ Calorie.}$$

$$1 \text{ Joule} = 6.241509479607718\text{e+18. eV}$$

[B] **Orange** = 1.1665 unit

[Bar section = 1]

$$1.1665 \text{ Calorie} = 3.048292187877699\text{e+19 eV.}$$

$$1.1665 \text{ Calorie} = 4.8839022 \text{ Joule.}$$

[Frequency & Energy section = 2]

$$1.1665 \text{ eV} = 4.463883926256329\text{e-20 Cal. } > \text{ This is ORANGE Frequency.}$$

$$1.1665 \text{ eV} = 1.868938922245\text{e-19 J.}$$

[Quantum section = 3]

$$1.1665 \text{ Joule} = 0.278613738415974 \text{ Cal.}$$

$$1.1665 \text{ Joule} = 7.280720807962403\text{e+18 eV}$$

[C] **Yellow** = 1.3331 Unit

[Bar section = 1]

$$1.3331 \text{ Cal} = 3.483650506352131\text{e+19 eV.}$$

$$1.3331 \text{ Cal} = 5.58142308 \text{ J.}$$

[Frequency & Energy section = 2]

$$1.3331 \text{ eV} = 5.101417627168721\text{e-20 Cal. } > \text{ This is YELLOW Frequency.}$$

$$1.3331 \text{ eV} = 2.135861532143\text{e-19 J.}$$

[Quantum section = 3]

1.3331 J = 0.3184054647941148 Cal

1.3331 J = 8.320556287265049e+18 eV

T₂ – Group

[D] **Green** = 1.4164 unit

[Bar section = 1]

1.4164 Cal = 3.701329665589347e+19 eV.

1.4164 Cal = 5.93018352 J

[Frequency & Energy section = 2]

1.4164 eV = 5.420184477624916e-20 Cal. > This is GREEN Frequency.

1.4164 eV = 2.269322837092e-19 J.

[Quantum section = 3]

1.4164 J. = 0.3383013279831852 Cal.

1.4164 J = 8.840474026916372e+18 eV.

[E] **Cyan** = 1.4997 Unit.

[Bar section = 1]

1.4997 Cal = 3.919008824826563e+19 eV.

1.4997 Cal = 6.27894396 J.

[Frequency & Energy section = 2]

1.4997 eV = 5.738951328081112e-20 Cal. > This is CYAN Frequency.

1.4997 eV = 2.402784142041e-19 J.

[Quantum section = 3]

1.4997 J = 0.3581971911722557 Cal

1.4997 J = 9.360391766567695e+18 eV.

[F] **Blue** = 1.5831 unit

[Bar section = 1]

** 1.5831 Cal = 4.136949303582671e+19 eV.

** 1.5831 Cal = 6.62812308 J.

[Here we can see that 6.62812308 unit, and 4.136930358267 unit all most SAME of Plunk constant, Next calculation is Done by as per this view.]

[Frequency & Energy section = 2]

1.5831 eV = 6.058100851827171e-20 Cal. > This is BLUE Frequency.

1.5831 eV = 2.536405664643e-19 J

[Quantum section = 3]

1.5831 J. = 0.3781169389509888 Cal.

1.5831 J. = 9.880933657166979e+18 eV.

[T3 Group: Violet [G] 1.9997 Cal. = 5.225606419287642e+19 eV.

1.9997 Cal. = 8.37234396 Joule. Bar section = 1, all are same calculation for sec.]

[** As per this Calculation, my view 1.5831 Cal = 6.62812308 J, 1.5831 Cal = 4.136949303582671e+19 eV. it is almost Same Digit of Plank Constant (6.626070040..x 10⁻³⁴ Js) & (4.135667662..x 10⁻¹⁵ eV.s)]

** So Here is my sree Debasish Dasgupta View that each and every Color have own separate Unit of (Bar's, Like **A, B, C, D, E, F, G,**) as per this calculation, to create Mass to light & light to color. Color to mass.

Example :

Red : 1 Calorie = 2.61319518892216e+19. eV

1 Calorie = 4.1868 Joule, (= '**A**' Bar in place of 'h' Bar)

Orange : 1.1665 Calorie = 3.048292187877699e+19 eV.

1.1665 Calorie = 4.8839022 Joule. (= '**B**' Bar In place of 'h' Bar)

Yellow : 1.3331 Cal = 3.483650506352131e+19 eV.

1.3331 Cal = 5.58142308 J. (= '**C**' Bar In place of 'h' Bar)

Green : 1.4164 Cal = 3.701329665589347e+19 eV.

1.4164 Cal = 5.93018352 J (= '**D**' Bar in place of 'h' Bar)

Cyan : 1.4997 Cal = 3.919008824826563e+19 eV.

1.4997 Cal = 6.27894396 J. (= '**E**' Bar in place of 'h' Bar)

Blue : 1.5831 Cal = 4.136949303582671e+19 eV.

1.5831 Cal = 6.62812308 J. (= '**F**' Bar in place of 'h' Bar)

[T3 Group: Violet 1.9997 Cal. = 5.225606419287642e+19 eV.

1.9997 Cal. = 8.37234396 Joule. (= '**G**' Bar in place of 'h' Bar]

Mathematical Example :

Question:- Electro light action of starting wave length is 7000 Angstrom to any metal transform it to the W₀ to Electron Volt ?. [h = 6.62 x 10⁻²⁷ erg-s, 1 eV = 1.6 x 10⁻¹² erg.]

Ans; W₀ = hv₀ = h(c/wave length₀) = (6.62 x 10⁻²⁷ x 3 x 10¹⁰) / (7000 x 10⁻⁸) erg,
= (6.62 x 10⁻²⁷ x 3 x 10¹⁰) / (7000 x 10⁻⁸ x 1.6 x 10⁻¹²) = 1.77 Electron Volt.

[If same process ;

Calculate Plank constant in that place (Blue, 1.5831 cal) = 6.62812308 unit (Here J.)

Example For Red : 1 Calorie = 4.1868 Joule, Here h = ('**A**' Bar) 4.1868 x 10⁻²⁷ erg-s And wave length = 8000, 1 eV = 1.6 x 10⁻¹² erg ,(Same Question)

Ans; W₀ = hv₀ = h(c/wave length₀) = (4.1868 x 10⁻²⁷ x 3 x 10¹⁰) / (8000 x 10⁻⁸) erg,
= (4.1868 x 10⁻²⁷ x 3 x 10¹⁰) / (8000 x 10⁻⁸ x 1.6 x 10⁻¹²) = 0.98128125 Electron Volt.

Ans ; eV = 0.98128125,

[0.98128125 / 0.413334 (L) = 2.3740637 (See Details about 2.37 in past article)

0.98128125 Calorie = 2.564276174⁺¹⁹ eV; $\sqrt{2.564276174} = 1.6013357$ unit

Energy, 1.6013357 x (10/9) = 1.7792619 Unit Mass, As per E = m-(m/10).

$\sqrt{1.7792619} = 1.3338 = 4/3Pj]$

For Orange: 1.1665 Calorie = 4.8839022 Joule.

Hare h = 4.8839022 x 10⁻²⁷ erg-S, ['**B**' Bar] And wave length= 6.858,A 1 eV = 1.6 x 10⁻¹² erg

Ans: 1.335275 eV. (Near 4/3)

[$\sqrt{1.335275} = 1.1555409.., 1.1555409x(10/9) = 1.28393444..x7 = 8.9875411..(C^2)$

1.335275/0.413334(L) = 3.23049882..., $\sqrt{3.23049882} = 1.7973588$ Unit]

For Yellow : $1.3331 \text{ Cal} = 5.58142308 \text{ J}$.
 $h = 5.58142308 \times 10^{-27} \text{ erg-S}$, ('C' Bar) And wave length = 5.999 A , $1 \text{ eV} = 1.6 \times 10^{-12} \text{ erg}$
Ans: $1.8382510876 \text{ unit}$,
 $[\sqrt{1.8382510876} = *1.35582118573(\text{Mag.Field/wave}), x, (10/9), = 1.506467984(\text{Energy})$,
 $*1.506467984 \times (10/9) = 1.6738533157 \text{ Unit mass}$.
 $1.3558211 \times 7 = 9.4907483$, $\sqrt[9]{9.4907483} = 1.284.. \times 7 = 8.98849 \text{ (C}^2 \text{)}$]

[N.b: * As per My calculation **Mag.Field or Wave x 10/9 or (M/m)² = Energy**,
Energy x (10/9) = Mass]

For Green: $1.4164 \text{ Cal} = 5.93018352 \text{ J}$, W/L = 5646 A
 $h = 5.93018352 \times 10^{-27} \text{ erg-S}$, ('D' Bar), W/L = 5646 A , $1 \text{ eV} = 1.6 \times 10^{-12} \text{ erg}$
Ans: 1.9693755 eV . $1.9693755 \times (9/10) = 1.7724379543$,
 $[\sqrt{1.7724379543} = 1.3313 = (4/3)$, $1.7724379543^2 = 3.14536 = \text{Pi}$.]

Same Used (E' Bar) Cyan = 2.1341802794 eV . Used (F' Bar) Blue = $2.331216894299 \text{ eV}$.
Violet ('G') Bar. = 3.92453623125 eV .

2.2 1st Method: Source of Quantum Digit.

(Mass Unit Equivalent JOULE value To CALORIE value) x Pi x 4 =
Quantum Digit of Light Frequency.

[source and how it is Inter-related with each ether. The quantum digits, Red :3, Orange: 3.5, Yellow: 4, Green: 4.25. Cyan: 4.5, Blue: 4.75; Reference :
Atomic Mass Energy Primary Colour, A Story Of Latest Scientific Discovery]

Example : (Same For All Colors) BLUE Mass Unit 1.5831,
 $1.5831 \text{ JOULE} = 0.3781169389509888 \text{ CALORIE}$. (Section 3)
 $0.3781169389509888 \times 3.1415926535897932384626433832795 \times 4 =$
 $4.7515575904251470204270868302612 \text{ Unit}$
(4.75 unit) of Quantum. Digit of Blue Color.

So: Calorie (sec.3) x 4Pi = Quantum of Visible Light.

[note: $0.3781169389509888 \times 4 = 1.5124677558039552$,
It is Energy, as per this calculation; [Rule $E = m - (m/10)$]
 $1.5124677558039552 \times (10/9) = 1.6805197286710613$.Unit Mass Value of particle
 $\sqrt{1.6805} = 1.296341 \text{ Approx}$. Same of Time x Space (3.6×0.36) = 1.296 .]
SO: CALORIE (sec.3) x 4 = ENERGY.
 $0.3781169389509888 \times \text{Pi} = 1.1878893976062867551067717075653$
 $1.1878893976062867551067717075653^3 = 1.6762084215224066959731 \text{ Mass}$.
SO: (CALORIE x Pi)³ = Mass.]

Yellow Mass unit = 1.3331 (example)

[Quantum section = 3]

$1.3331 \text{ J} = 0.3184054647941148 \text{ Cal}$

$0.3184054647941148 \times 4\text{Pi} = 4$; 4 is quantum Digit of yellow. (Like This for all)

Next:: Calculation found some of new rule.

Frequency (or calorie of section 2) / Calorie (section.3) = 10 Energy

Example : This result is same for all.

Green = 1.4164 unit

1.4164 eV = 5.420184477624916e-20 Cal. > (Section 2)
1.4164 J = 0.3383013279831852 Cal. > (Section 3)

5.420184477624916 / 0.3383013279831852 = **16.0217653..It is 10 times of ENERGY.**
16.0217653./ 9 = 1.7802 Unit Mass.

Orange: 1.1665 unit

[Frequency & Energy section = 2]

1.1665 eV = **4.463883926256329e-20 Cal.**

[Quantum section = 3]

1.1665 Joule = 0.278613738415974 Cal.

4.463883926256329 / 0.278613738415974 = 16.02176529999999928610053..Unit.

It's 10 times of ENERGY. 16.0217653../9 = 1.7802 Unit Mass. (Same result For all)

4.463883926256329e-20 / 0.278613738415974 = 1.6021765299999999286100530681526e-19, (eng.)

[When Create one Color From one Mass in that Moment it is Flexible (Quantum Digit, Constant, etc.)
Frequency, It is depends on Particle's Quality; Mass, etc.)

For..Exap:4.7515575904251470204270868302612.x.1.2919(Constant)=

6.1385372510702474356897534760144 Frequency of BLUE Light. All most Same of the Past article
Chart/Table]

Example For ALL Colors : (Same result.)

Yellow = 1.3331 Unit

1.3331 Calorie = 5.58142308 Joule. (Section 1, Bar)

1.3331 Joule = 0.3184054647941148 Calorie. (Section 3)

5.58142308 / 0.3184054647941148 = 17.52929424..., $\sqrt{17.5292942} = 4.18688 = 4/3 \text{ Pi}$.

Blue = 1.5831 unit

1.5831 Cal = 6.62812308 J. (Bar.Section.1)

[Quantum section = 3]

1.5831 J. = 0.3781169389509888 Cal.

$\sqrt{6.62812308 / 0.3781169389509888} = 17.529294240] = 4/3 \text{ Pi}$.

SO JOULE (Sec.1) / Calorie (sec.3) = $(\frac{4}{3} \text{ Pi})^2$ (Rule)

[Calorie (sec.3) / Joule (sec.1) = 2.38845, or Frequency / Joule = 2.38845 unit.

Example: For All Colors. Same. Result.

When Red Mass 1 unit.

1 eV = 1.60217653e-19 Joule (section 2)

1 ev = 3.826732898633801e-20 .Calorie. (Section 2)

3.826732898633801 / 1.60217653 = 2.3884589662749591020410216594547 Unit.

2.38845³ = 13.62 Bohr 'H' Atom Energy unit.

$\sqrt[3]{2.38845} = 1.3367 = 4/3$, $1.3367^2 = 1.786755$ Unit Mass.

$\sqrt{2.38845} = 1.5454356$ unit Energy, $\sqrt{1.5454356} = 1.2431$ (Max. Photon Energy)]

**# CALORIE (sec.2) / eV (sec.1) or (Frequency / eV) = Magnetic Field Either Wave,
Magnetic Field Either Wave Cube = Pi.** (Without any power use digit only)

Example: **For All Colors. Same. Result.**

Cyan = 1.4997 Unit.

1.4997 Cal = 3.919008824826563e+19 eV. (Section 1)

1.4997 eV = 5.738951328081112e-20 Cal. > This is CYAN Frequency (Sec.2)

5.738951328081112 / 3.919008824826563 = 1.464388467748625481260792608..

1.4643884677486³ = **3.1402838** = Pi. 1.4643884677486 x (10/9) = 1.6270983 Unit

eV(Section 2) / Joule (Sec.3) = Red Frequency

Example: For All Colors. Same. Result.

Blue = 1.5831 unit

1.5831 eV = 2.536405664643e-19 J. (Section 2)

1.5831 J. = 9.880933657166979e+18 eV. (Section 3)

9.880933657166979 / 2.536405664643 = **3.895644** ...Unit (Red Frequency digit)

3.895644 / 3 = **1.298548** Unit it is Constant. (See Later)

1.298548² = **1.68622695886366**.. unit is Mass.

3.895644² = 15.176042174736, 15.176042174736 / 9 = **1.6862269**..Same.

1.298548 x 4 = 5.194192, [Will be discuss in following chapter (root 3)³]

1.29858/0.413334 (L) = 3.14172073916..= ONE Pi.

$\sqrt[3]{5.194192} = 1.731832955424 = \text{Root 3}$, (See Later)

3.895644... / (1.2985...)³ = **1.78..Unit Mass** (Mass Before create a Nucleus),

(4 Pi x 1.24 unit Photon eV.) / 9 = $\sqrt{3}$

Either [(4 x 3.141592653589793..) x 1.24] = 15.582299561805374462774711181066

15.582299561805374462774711181066 / 9 = 1.7313666179783749403083012423407

(1.7313666179783749403083012423407)⁴ = **8.9857878102652620998185830876655**

= **Speed of light Square.**

4 x 1.24 (Photon eV.) = 4.96, $\sqrt[3]{4.96} = 1.705403796656319538384519..x 2 =$

3.410807593312639.., $\sqrt[3]{3.410807593312639} =$

1.5052861776391507101086348848065 x (10/9) = **1.6725402 unit (mass)]**

SO: This calculation either 3.89..digit/unit present in all Colors And Particle.

3.895644... is **RED** Frequency,

[1 Calorie = 4.1868 Joule, 4.1868 = 4/3 Pi. Or 4.1868 / (4/3) = Pi.]

Red is Basic / Base of Particle to Color either Color to Particle. I have discuss several time in past ARTICLE. $\sqrt[3]{4.18688} = (1.611746879..) = 2.597728$, $2.597728 / 2 = 1.2988$, It is Constant. $(1.2988)^2 = 1.687$ mass. will be discuss following chapter .]

2.3

2nd Method 4 Calorie x 1.2985 (Constant) = Quantum Digit

Either [$(\sqrt{3})^3$ x Calorie,] = Quantum Digit. [4 x 1.2985.., = 5.194.., = [$(\sqrt{3})^3$]

This Calculation Done By Each Color's Frequency's JOULE to Equivalent Electron Volt (without Power, used only Digit) to Joule's Digit.

Red frequency. = 3.8757 ,

3.8757 Joule equivalent eV = **2.419021829011563 eV.**(x 10¹⁹)

2.419021829011563 x 0.413334 = 1 unit,

Orange frequency.= 4.5216, 4.5216 Joule = **2.822160926299426 eV.**
2.822160926299426 x 0.413334 = 1.1665

Yellow frequency.= 5.1676, 5.1676 joule = **3.22536243868208 eV**
3.22536243868208 x 0.413334 = 1.3331

Green frequency.= 5.4905, 5.4905 joule = **3.426900779778618 eV**
3.426900779778618 x 0.413334 = 1.4164

Cyan frequency.= 5.8135, 5.8135 joule = **3.628501535969947 eV**
3.628501535969947 x 0.413334 = 1.4997

Blue frequency.= 6.1365, 6.1365 Joule = **3.830102292161276 eV**
3.830102292161276 x 0.413334 = 1.5831]

Example for **Red frequency.**= 3.8757 ,
3.8757 Joule equivalent eV = 2.419021829011563 eV. (x 10¹⁹)
2.419021829011563 Joule = 0.5777734525547674 Cal

Example: **Red** : 2.419021829011563 Joule = 0.5777734525547674 Calorie.
0.5777734525547674 x 4 x 1.2985 = **3** it is **Quantum Digit of Red Frequency**

. [note 2.419021829011563 Joule = 1.509834806503064e+19 eV. 1.509834806503064e+19 x (10/9) = **1.6775942294478488..89e+19** mass.
(3.895644 / 3 = 1.298548, Unit it is Constant. And Pi x 0.413334 = 1.2985 Already Explain about it.
L = 0.413334 see about it in past article), & (1.298548)² = 1.6862269 / 10 = 0.1686226908304,
 $\sqrt[3]{0.1686226908304} = 0.552465723615369431..$ Unit, ;(Another used 'e' mass.), 0.5524657236153694 x 3
= **1.65739717084610829...** $\sqrt{1.6573971708461} = \underline{\mathbf{1.2873993828}}$, 1.28739938280476894...; x 7 =
9.01179567..]

Yellow frequency. = 5.1676,
5.1676 joule = 3.22536243868208 eV

3.22536243868208 Joule = 0.7703645836156683 Cal.
3.22536243868208 Joule = 2.013113023620487e+19 eV.

Same Calculation For All Colors.

4 Calorie x 1.2985 (Constant) = Quantum Digit or Calorie x ($\sqrt[3]{3}$)³

Example: **Yellow:** 3.22536243868208 Joule = 0.7703645836156683 Calorie
0.7703645836156683 x 4 x 1.2985 = **4** It is **Quantum Digit of Yellow Frequency.**

Like This for All Colors. [As per mention Result proved that Used Digit in Present and previous article Mathematically True. For details; explain with example by Images. (Below)]

Internal discussion: about this result

RED: 2.419021829011563 eV = 0.9256950653403168e-19 Cal.

0.9256950653403168 x 4 x 1.2985 = 4.8 It is Another use Unit For Energy Either Charge,
So Red Color Energy = 4.8 Unit. (**Charge on an electron = 4.8 x 10⁻¹⁹ stat C**)

[Used Visible Light's energy Chart:-- Example:
Red 2.419021829011563 Joule = **1.509834806503064e+19** eV.

1.509834806503064 x (10/9) = 1.677594229447848..Unit of Mass.
As per 'E=m-(m/10),

$$(1.509834806503064)^4 = \mathbf{5.196581370839},$$
$$\sqrt[3]{5.196581370839} = 1.732098467 = \mathbf{Root\ 3}.$$

(Why Here use to the power 4 and Cube root, Because we know that Sphere rule $4/3 \text{ Pi} \times r^3$ cube, So used 4 and 3 = cube.),

[1 Calorie = 4.1868 Joule, 4.1868 = $4/3 \text{ Pi}$. Or $4.1868 / (4/3) = \text{Pi}$.]

1.509834806503064 x 9 = 13.588 it is Bohr Energy.

$$\sqrt[3]{1.509834806503064} = 1.04684178 = 1/3 \text{ Pi. Or } 1.04684178 \times 3 = \text{Pi}.$$

Red is Basic / Base of Particle to Color either Color to Particle. I have discuss several time in past ARTICLE, I think it is enough.]

JOULE / CALORIE = 4/3 Pi. Or Frequency / Calorie = 4/3 pi

Example: Same process For All Colors.

RED : 2.419021829011563 eV = 0.9256950653403168e-19 Cal
2.419021829011563 eV = 3.875700099566838e-19 Joule
 $3.875700099566838 / 0.9256950653403168 = \mathbf{4.1868 = 4/3 \text{ Pi}}$.

OTHERS Example : Same for all.

RED: 2.419021829011563 eV = 3.875700099566838e-19 Joule
2.419021829011563 Joule = 0.5777734525547674 Cal

3.875700099566838 (Red Fqn.) / $0.5777734525547674 = 6.708$.

$$\sqrt[4]{6.708} = 1.60934, 1.60934 \text{ Cube} = 4.16816 \text{ (Nearer } 4/3 \text{ Pi)}$$

1.60934 (Energy) x (10/9) = 1.788 Unit Mass. $(1.60934)^2 = 2.5899752356, 2.5899752356 / 2 = 1.295$
(Nearer Time x Space $3.6 \times 0.36 = 1.296$).

$1.295^2 = \mathbf{1.67699 \text{ Unit mass}}$. $\sqrt[3]{6.708} = 1.8859, (1.8859)^4 = 12.649 = 4 \text{ Pi. (Approx)}$.

$12.65 / 4 = 3.1625, (3.1625)^2 = 10. \sqrt{3.1625} = \mathbf{1.77834 \text{ Unit Mass}}$

[Again : $3.875700099566838e-19 \times 0.5777734525547674 = 2.2392766275935877634012118434812e-19 \times (3/4) = \mathbf{1.6794574706951908225509088826109e-19 \text{ (Unit mass)}}$

$(3/4) = 0.75, (0.75)^3 = 0.421875, 0.421875 \times 4 = \mathbf{1.6875 \text{ (Unit mass)}}$,

$0.5625 \times (\sqrt{10}) = \mathbf{1.6649392 \text{ (Unit Mass)}}$

$1.6875 / 3 = 0.5625, 0.5625 \times 3.1625 (= 1.77890625 \text{ or } \mathbf{1.78... \text{ (Unit Mass.)}}$

$(0.5625)^3 = \mathbf{0.177978515625}$. or 0.178 , So $1.78 - 0.178 = \mathbf{1.602 \text{ unit energy}}$.

$0.5625 \times 2 = 1.125 \times 10 = 11.25$ it minimum angel of one electron's. see about past article.

$(1.125)^4 = \mathbf{1.6018 \text{ It is Energy}}$.

$(5625)^3 = \mathbf{0.177978515625}$. or 0.178 , So $1.78 - 0.178 = \mathbf{1.602 \text{ unit energy.]}$

Next. Calculation Same for all

Green Frequency. = 5.4905, 5.4905 joule = 3.426900779778618e+19
3.426900779778618 Calorie = 14.34774818477712 Joule

3.426900779778618 Calorie = 8.955160630631081e+19 ev.
3.426900779778618 eV = 1.311383395433267e-19 Calorie.

14.34774818477712 / 1.311383395433267 = 10.94092561 Unit.
 $\sqrt{10.94092561} = 3.3$ it is Displacement, (s) see E = F/s in past article.

8.955160630631081 / 1.311383395433267 = **6.828789..Unit.**
6.828789 x2 = 13.657578 Unit (Bohr Energy)

6.828789 x 4 =27.315156, 27.315156 / 3 = 9.105052 =(4/3) x 6.828789= Unit (Another Used unit Mass)
 $\sqrt[3]{27.315156} = 1.44411$, it is Mag. Filed Either Wave,
1.44411 x (10/9) = **1.6045671..Unit Energy**, 1.6045671 x (10/9)=**1.7828524 Unit Mass**,

[We know Total 50 no. of electron stay in four orbit (32, 8, 8, 2), so 360 degree for one orbit, **4 x 360 = 1440** Degree total acquire place, as per this average for one electron 1440 / 50 = 28.8, 28.8 / 2 = 14.4, 14.4 x 9 = **1.6** unit, Mass either energy for one part.]

6.828789 / 2 = 3.4143945, $\sqrt[3]{3.4143945} = 1.50581366..$ (Energy), 1.50581366 x (10/9) = **1.6731263 Unit.**
(mass)

6.828789 / 3 = 2.276263, $\sqrt{2.276263} = 1.5087289352299..$ Energy 1.5087289352299..x (10/9) =
1.676365483588797 Unit Mass.

6.828789 / 6 = 1.1381315, (1.1381315)² = 1.29534...(Time x Space)
1.295343311² = **1.6779142941..Unit Mass.**

3.1

Hence : Found average 'mass' value (like 1.67..,1.688..;1.679..;) and Mass value (like 1.777..; 1.778..; 1.78..; see Details by images); by ensuring this result; Make's new rule:

As per previous articles Force = 5.334 [ref. e = f/s (s= displacement).]

Another proof for, F = 5.334, as per Law of Magnetic Charge in a Magnetic Field "F = qv x B",:: F = Force, q = Charge, B = Magnetic field (in a particular point), v = velocity,

As per present and past Calculation, 'q' = 1.2345678, unit. is charge (eV. of Photon, it will be again discussed in following chapters. In images). 'v' = 3 unit, B = 1.44...; unit. So. F = [(1.2345678 x 3) x 1.44..;] = 5.3333; Unit.

Either: [Magnetic Field = $\sqrt[3]{3}$ = 1.44224957...]& Charge = 1.602
As per: (1.44224957)² x (1.602)² = 5.338 = FORCE,

[note: Each step ; (1.44224957)² x 1.602 = 3.33..; Displacement (s),
1.44224957 x 1.602 = 2.31048, it is adding position Time & Place, discussed in past article.
Or, 2.31048² = 5.3383178304 Or, 1.44224...;eV = 2.3107230786272e-19 J,

Conclusion:

- *1) $(7/22) / (3/4) = 'r' \text{ Cube [0.4242]}$
- *2) $\text{Root 3 / energy Cube} = 'r' \text{ Cube [0.4274]}$
- *3) $(\text{radius of 'H' Atom's either 'G'}) / \text{Photon eV.} = 'r' \text{ Cube [0.4274]}$
- *4) $1.29858 (\text{Constant}) / \text{Root 3} = 'r' [0.75]$
- *5) $3.895644 / 3 = \text{Constant Unit [1.298548]}$
- *6) $\text{Pi} \times 0.413334 = \text{Constant Unit [1.2985]}$
- *7) $\text{Joule / Calorie} = (4/3) \text{ Pi. [4.1868]}$
- *8) $1.29858 (\text{Constant}) / 0.413334 = \text{Pi}$
- *9) $1.2985848 (\text{Constant}) \times 4 = (\text{Root 3}) \text{ Cube.}$
- *10) $1.29858 (\text{Constant}) / (3/4) = \text{Root 3}$
- *11) $\text{Displacement} \times \text{Mass} \times ('r' \text{ of 'H' Atoms either 'G'}) = \text{Pi.}$
- *12) $(4L / \text{Photon eV}) \times (\text{Displacement} \times 'G' \times \text{mass}) \times ('r' \text{ of 'H' Atoms} / \text{Photon eV}) = \text{mass}$
- *13) $(\text{Joule / Calorie}) \times \text{Calorie (Mass equivalent)} = \text{Mass}$
- *14) $\text{Cube root over (electron Volt / Joule)} = \text{Power Factor (0.8546.;)}$
- *15) $\text{9root over (Calorie / Joule)} = \text{Power Factor (0.8529...)}$
- *16) $\text{9root over } [(7/22) \times (3/4)] = 0.8528244...; (\text{Power Factor})$
- *17) $\text{Force} = (\text{Mass} \times \text{mass}) / \text{radius square [F = (M} \times \text{m) / r}^2 \text{]}$
And, $(\text{speed of light}) / (3/4)^2$
- *18) 1.2857142857142857 Is a Constant Digit
- *19) $\text{Force} = [\text{Displacement} \times \text{Magnetic field.} \times \text{Root over (photon) eV.}]$
- *20) Every Particle has three different sections in it. In every section; the unit of (mention), resultant Bar & others Sections (A; B; C; D; E; F; G) like 'Planck's constant is incumbent. The efficiency of the digits (4/3), (Pi), ($\sqrt{3}$); (3), (3.89..), (9), (10), (0.413334) etc, etc, is definitely present in those digits

Reference:

***a) “Atomic Mass Energy Primary Colour”**

***b) “Atom to Nucleus”**

***c) “ Atomic Time Space”**

***d) “A Story Of Latest Scientific Discovery”**

Site : Mendeley Data, Datacite; Narcis.nl. All articles and link also available in great social media Blog, Twitter, Facebook, By Google Search

Shown present unit (Calorie; Joule; eV, etc.) How to be compared. The Digit or Numbers are calculated

Image for example (1) 'Atomic Mass and Constant' see Debasish Dasgupta

$\sqrt[9]{9}$ or $\sqrt[3]{3} = 1.44224957$ (Joule) = $9 \times (10^{-18} \text{ eV})$; $1.44224957 \times (10/9) = 1.602499$ Unit Energy ($E_{...1}$), $1.602499 \times (10/9) = 1.7805555$ Unit (Mass Before create a Nucleus), 1.7805555 Mass... ($M_{...1}$)

$1.44224957 \text{ eV} = 2.31 \times (10^{-19}) \text{ Joule}$. $2.31^2 \text{ square} = 5.3361$ Unit ($Force_{...1}$) $\sqrt{2.31} = 1.5198684...$ $\times (10/9) = 1.68874268373007373685208268956$, Unit mass. ($m_{...1}$)

$[5.3361 / 3 = 1.7787, 1.7787^2 = 3.16377369 \text{ Unit (f or acceleration)}]$; $3.16377369^2 = 10$; $\leftrightarrow 5.3361 \times 3 = 16.0083$; $16.0083 / 9 = 1.7787$; $\sqrt{1.7787} = 1.333679 = (4/3 \dots 1)$;

$5.3361 \times (10) = 53.361$; $53.361 / 9 = 5.929$; $\sqrt[4]{5.929} = 1.56043384939...$ $\times (10/9) = 1.73381$; $1.73381^2 = 3$; $(1.7787, M_{...2}; 5.3361 \text{ Force } F_{...1})$

$\sqrt[4]{5.3361} = 1.519868415357...$ $\times (10/9) = 1.68874268373$. Unit ($m_{...2}$) (After create a Nucleus), $\sqrt{1.68874...} = 1.299$ (Constant... $Con_{...1}$) (Atom to Nucleus' $E \times M = m^2$)

$1.78 \text{ eV} = 2.8518742234e-19 \text{ Joule}$, $\sqrt{2.8518742234} = 1.68874$. Unit (After create a Nucleus), $1.78 \text{ Joule} = 1.110988687370174e+19 \text{ eV}$, (almost Same Mass 1,2...; mass 1,2...)

$1.110988687370174 = (10/9)$, $\sqrt{1.11098868} = 1.05403448...$ (M / m), (Please see Past article Atom to Nucleus) $\sqrt[3]{3} \times (0.9) = 1.298...$; (Constant); $\sqrt{3} \times (3/4) = 1.299$ Constant

$\sqrt[3]{3} / (3/4) = 1.601249273568$; $1.601249273568 \times (3/4) = 1.200936955176$; $1.200936955176^2 = 1.4422495703...$; $1.4422495703...$; $\times (10/9) = 1.60249952$ (energy)

$\sqrt[3]{3} = 1.4422495703...$; $1.4422495703...$; $1.4422495703...$; $(7/22) = 4.53278436...$; Cube Root $4.532784363...$; $= 1.654963245086...$; Root $1.654963245086...$; $= 1.28645374774...$; $\times 7 = 9'$

$1.4422495...$; $\times (10/9) = 1.6024995$ (eng) $\times (10/9) = 1.78055502...$ (M) [$1.44224957...$; $\times 9 = 12.980246...$; Cube Root $12.980246...$; $= 2.350143...$; Root $2.350143 = 1.533017...$; $\times (10/9) = 1.70335...$]

[We Know One Neutron Mass = $1.6749 \times 10^{-27} \text{ Kg}$ & Proton mass = $1.6726219 \times 10^{-27} \text{ Kg}$, enrgy = $1.60217653 \times 10^{-19} \text{ J}$, $1.60217653 \times 3 = 4.806...$; Cube rt $4.806 = 1.68763$ (m)]

$(7/22) / (3/4) = 0.424242...$ = 'r' Cube. ('r' Cube...1) (As per $4/3 \text{ Pi } 'r'$ Cube), So $4.18879 \times 0.424242 = 1.777...$ Unit Mass, (M); $0.424242 \times 4 = 1.696968$ unit mass. (m).

$(3/4) / (7/22) = 2.357142857...$; $\sqrt{2.357142857...} = 1.5352989471...$; $\sqrt{1.5352989471...} = 1.239...$; (eV) [$1.5352989471...$; $\times (10/9) = 1.7058877...$]; $\sqrt[3]{2.357142857...} = 1.3308$ (...4/3) $1.70588771906...$; $\times 2 = 3.411775438...$; Cube Root $3.4117754381...$; $= 1.5054285433...$; $1.5054285433...$; $\times (10/9) = 1.67269838...$; (m) ■ Proton mass = $1.6726219 \times 10^{-27} \text{ Kg}$]

That means When create one Mass to Nucleus mass in that moment Present, Force = 5.3361 Unit, Magnetic Field Either Wave = 1.44224957, f or acceleration = 3.16377369 & Energy = 1.602499 Unit. Time for acceleration = 1.11098868 when 'v' = 3, (Reference: 'Atomic Mass Energy Primary Colour' A Story Of Latest Scientific Discovery')

Now ; This Calculation Done By Each Colors Frequency's Equivalent Mass to Calorie, eV, Joule,

[Let us see by Identical Result; if we convert each Frequency Joule to Electron Volt then - DIFFERENT type of SIX CONSTANT

(This calculation /Chart /Table Done in past article "Atomic Mass Energy Primary Colours" & "A Story of Latest Scientific Discovery" Please see.)]

TABLE: 1

Red fqn. = 3.8757 , 3.8757 Joule equivalent eV = 2.419021829011563 ($\times 10^{-19}$) eV. 2.419021829011563 $\times 0.413334 = 1$ unit.
Orange fqn. = 4.5216, 4.5216 Joule = 2.822160926299426 ($\times 10^{-19}$) eV. 2.822160926299426 $\times 0.413334 = 1.1665$
Yellow fqn. = 5.1676, 5.1676 joule = 3.22536243868208 ($\times 10^{-19}$) eV 3.22536243868208 $\times 0.413334 = 1.3331$
Green fqn. = 5.4905, 5.4905 joule = 3.426900779778618 ($\times 10^{-19}$) eV 3.426900779778618 $\times 0.413334 = 1.4164$
Cyan fqn. = 5.8135, 5.8135 joule = 3.628501535969947 ($\times 10^{-19}$) eV 3.628501535969947 $\times 0.413334 = 1.4997$
Blue fqn. = 6.1365, 6.1365 Joule = 3.830102292161276 ($\times 10^{-19}$) eV 3.830102292161276 $\times 0.413334 = 1.5831$
Total -- 7.9988 + 1 = 8.9988 or 9, (Red = 1 unit, one another Red present See Past articles.) [T 3 Group Violet fqn. = 7.7514, 7.7514 joule = 4.838043658023127 ($\times 10^{-19}$) eV 4.838043658023127 $\times 0.413334 = 1.9997$]

T₁ - Group (As per Table: 1 Calculation)

When Red Mass 1 unit.

[Bar section = 1]

1 Calorie = 2.61319518892216e+19. eV

1 Calorie = 4.1868 Joule, 4.1868 = $4/3 \text{ Pi}$. Or $4.1868 / (4/3) = \text{Pi} (4/3 \dots 1)$

[Frequency & Energy section = 2]

1 eV = 3.826732898633801e-20 Calorie. > This is RED Frequency.

1 eV = 1.60217653e-19 Joule

[Quantum section = 3]

1 Joule = 0.2388458966274959. Calorie.

1 Joule = 6.241509479607718e+18. eV

Orange = 1.1665 unit

[Bar section = 1]

1.1665 Calorie = 3.048292187877699e+19 eV.

1.1665 Calorie = 4.8839022 Joule.

[Frequency & Energy section = 2]

1.1665 eV = 4.463883926256329e-20 Cal. > This is ORANGE Frequency.

1.1665 eV = 1.868938922245e-19 J.

[Quantum section = 3]

1.1665 Joule = 0.278613738415974 Cal.

1.1665 Joule = 7.280720807962403e+18 eV

Yellow = 1.3331 Unit

[Bar section = 1]

1.3331 Cal = 3.483650506352131e+19 eV.

1.3331 Cal = 5.58142308 J.

[Frequency & Energy section = 2]

1.3331 eV = 5.101417627168721e-20 Cal. > This is YELLOW Frequency.

1.3331 eV = 2.135861532143e-19 J.

[Quantum section = 3]

1.3331 J = 0.3184054647941148 Cal

1.3331 J = 8.320556287265049e+18 eV

[T3 Group: Violet 1.9997 Cal. = 5.225606419287642e+19 eV. 1.9997 Cal. = 8.37234396 Joule. Bar section = 1, all are same calculation for sec.]

(2 Times of Red Color's Calculation)

TABLE: 2

Color	Quantum	Constant	Frequency	Wave L
Red	3	$\times 1.2919$	= 3.8757	8000 Unit
Orange	3.5	$\times 1.2919$	= 4.5216	6858 Unit
Yellow	4	$\times 1.2919$	= 5.1676	5999 Unit
Total	10.5		13.5649 Unit	20854 Unit
Total T ₁	13.5 (10 + 3.5)	(13.5649 + 3.87570)	17.44	28854 Unit
As per Calculation. Another Red Frequency is present in T1 Group so here add one Red Fqn. Quant. Const. etc. For details. Please see Past Article.				
Green	4.25	$\times 1.2919$	= 5.4905	5646 Unit
Cyan	4.5	$\times 1.2919$	= 5.8135	5332 Unit
Blue	4.75	$\times 1.2919$	= 6.1365	5052 Unit
Total	13.5		17.44	16030 Unit
Voilet	6	$\times 1.2919$	= 7.75	4000 Unit T ₂

T₂ - Group

Green = 1.4164 unit

[Bar section = 1]

1.4164 Cal = 3.701329665589347e+19 eV.

1.4164 Cal = 5.93018352 J

[Frequency & Energy section = 2]

1.4164 eV = 5.420184477624916e-20 Cal. > This is GREEN Frequency.

1.4164 eV = 2.269322837092e-19 J.

[Quantum section = 3]

1.4164 J = 0.3383013279831852 Cal.

1.4164 J = 8.840474026916372e+18 eV.

Cyan = 1.4997 Unit.

[Bar section = 1]

1.4997 Cal = 3.919008824826563e+19 eV.

1.4997 Cal = 6.27894396 J.

[Frequency & Energy section = 2]

1.4997 eV = 5.73895132808112e-20 Cal. > This is CYAN Frequency.

1.4997 eV = 2.402784142041e-19 J.

[Quantum section = 3]

1.4997 J = 0.3581971911722557 Cal

1.4997 J = 9.360391766567695e+18 eV.

Blue = 1.5831 unit

[Bar section = 1]

1.5831 Cal = 4.136949303582671e+19 eV. <

1.5831 Cal = 6.62812308 J. <

[Here we can see that 6.62812308 unit, and 4.136930358267 unit almost SAME DIGIT of Plunk constant, if accept it is same (without unit) Next calculation Done by as per this view.]

[Frequency & Energy section = 2]

1.5831 eV = 6.058100851827171e-20 Cal. > This is BLUE Frequency.

1.5831 eV = 2.536405664643e-19 J.

[Quantum section = 3]

1.5831 J = 0.3781169389509888 Cal.

1.5831 J = 9.880933657166979e+18 eV.

Here is comparing with 'Planck's constant from different method

Image for example (2) 'Atomic Mass and Constant' sree Debasish Dasgupta

[** As per this Calculation, my view it is Allmost Same to Plank Constant (6.626070040... x 10⁻³⁴ Js) and (4.135667662... x 10⁻¹⁵ eV.s). & Blue = 1.5831 Cal = 6.62812308 J. 1.5831 Cal = 4.136949303582671e+19 eV.]

** So Here is my sree Debasish Dasgupta View that each and every Color have own separate BAR Unit of calculation, for create Mass to light & light to color. Color to mass.

Table - 1 (a) [LIKE THIS: Bar section = 1 From Table: 1 Calculation]

Red : 1 Calorie = 2.61319518892216e+19 eV 1 Calorie = 4.1868 Joule, (= 'A' Bar in place of 'h' Bar)
Orange : 1.1665 Calorie = 3.0482921877699e+19 eV. 1.1665 Calorie = 4.8839022 Joule. (= 'B' Bar In place of 'h' Bar)
Yellow : 1.3331 Cal = 3.4836506352131e+19 eV. 1.3331 Cal = 5.58142308 J. (= 'C' Bar In place of 'h' Bar)
Green : 1.4164 Cal = 3.701329665589347e+19 eV. 1.4164 Cal = 5.93018352 J. (= 'D' Bar in place of 'h' Bar)
Cyan : 1.4997 Cal = 3.919008824826563e+19 eV. 1.4997 Cal = 6.27894396 J. (= 'E' Bar in place of 'h' Bar)
Blue : 1.5831 Cal = 4.136949303582671e+19 eV. 1.5831 Cal = 6.62812308 J. (= 'F' Bar in place of 'h' Bar)
[T3 Group: Violet 1.9997 Cal. = 5.225606419287642e+19 eV. 1.9997 Cal. = 8.37234396 Joule. (= 'G' Bar in place of 'h' Bar)]

Total Joule: As per Table 1 (a), 4.1868 + 4.8839022 + 5.58142308 = 16.65212528 (T1)
5.93018352 + 6.27894396 + 6.62812308 = 18.83725056 (T2), [8.37234396 T3]
Internal step: (T1) > 16.65212528 / 3 = 4.88404176 (avg). Cube root 4.88404176 = 1.69665336536, 4root > 4.88404176, = 1.486602443 x (10/9) = 1.651780492206,
root > 1.651780492206 = 1.2852161...x 7 = 8.9985,
(T2), 18.83725056 / 3 = 6.27908352, Cube root > 6.27908352 = 1.844868518329169
1.844868518329169, x (9/10) = 1.660381666496,
T3 > 4 root > 8.37234396 = 1.701029187136386571679322503445
1.70102918713638657...x 2 = 3.402058374272773...,
cube rt > 3.402058374272773... = 1.504, 1.504 x (10/9) = 1.6711088
(T1) + (T2) = 33.48937636 / 6 = 5.5815627266...; Cube rt > 5.5815627266666 =
1.773857, < Sqr = 3.1465686149(Near Pi), 6Root > 33.48937636 = 1.7953584773403
Total (T1)+(T2)+(T3) = 41.86172032 (10 times of Red), 8Root > 41.86172 =
1.59487677571192709 x (10/9) = 1.7720853063466, < sqr = 3.1402863329694723.Pi
41.86172 / 9 = 4.6513022, Cube root > 4.6513022 = 1.669263327421860526451
Cube > 4.6513022 = 100.629...; 4Root > 100.629 = 3.1672395926 < Rt = 1.77967

Example : Electro light light action of starting weve length is 7000 Angstrom to any metal transform it to the W₀ to Electron Volt.

$$[h = 6.62 \times 10^{-27} \text{ erg-s, } 1 \text{ eV} = 1.6 \times 10^{-12} \text{ erg.}]$$

Ans; W₀ = hν₀ = h(c/wave length)₀ = (6.62 x 10⁻²⁷ x 3 x 10¹⁰) / (7000 x 10⁻⁸) erg,
= (6.62 x 10⁻²⁷ x 3 x 10¹⁰) / (7000 x 10⁻⁸ x 1.6 x 10⁻¹²) = 1.77 Electron Volt.

[If same process ; same calculation, Calculate Plank constant 'h' in that place use 'A'; 'B'; 'C' etc then...] (For Red 'A' Bar in place of 'h' Bar)

For Red : = 1 Calorie = 4.1868 Joule, Hare h = ('A' Bar) 4.1868 x 10⁻²⁷ erg-S And weve length = 8000, Angstrom 1 eV = 1.6 x 10⁻¹² erg. (Same Question)

Ans; W₀ = hν₀ = h(c/wave length)₀ = (4.1868 x 10⁻²⁷ x 3 x 10¹⁰) / (8000 x 10⁻⁸) erg,
= (4.1868 x 10⁻²⁷ x 3 x 10¹⁰) / (8000 x 10⁻⁸ x 1.6 x 10⁻¹²) = 0.98128125 Electron Volt.
Ans ; eV = 0.98128125, (Red electron Volt)

[0.98128125 / 0.413334 (L) = 2.3740637 [See Details about 2.37...(1) in past article]
0.98128125 Calorie = 2.564276174⁺¹⁹ eV √2.564276174 = 1.6013357 unit
Energy, 1.6013357 x (10/9) = 1.7792619 Unit Mass, As per E = m·(m/10).
√1.7792619 = 1.3338 = 4/3Pi]

For Orange: 1.1665 Calorie = 4.8839022 Joule.

, Hare h = 4.8839022 x 10⁻²⁷ erg-S, ['B' Bar] And wave length= 6.858, A 1 eV = 1.6 x 10⁻¹² erg

Ans: 1.335275 eV. (Near 4/3)

[√1.335275 = 1.1555409... 1.1555409 x (10/9) = 1.28393444...x7 = 8.9875411...=C²
1.335275/0.413334(L) = 3.23049882...√3.23049882 = 1.7973588 Unit]

For Yellow : 1.3331 Cal = 5.58142308 J.

.h= 5.58142308 x 10⁻²⁷ erg-S, ('C'Bar) And weve length = 5.999, A.

1 eV = 1.6 x 10⁻¹² erg

Ans: 1.8382510876 unit,

[Root, 1.8382510876 = 1.35582118573 (Mag.Field/weve), x₁(10/9), = 1.506467984 (Energy), 1.506467984 x (10/9) = 1.6738533157 Unit mass.
1.3558211 x 7 = 9.4907483, √9.4907483 = 1.284...x 7 = 8.98849 = C²]

[N.b: * As per My calculation Mag.Field or Wave x 10/9 or (M/m)² = Energy, Energy x (10/9) = Mass]

For Green: 1.4164 Cal = 5.93018352 J, W/L = 5646, A

.h= 5.93018352 x 10⁻²⁷ erg-S, ('D'Bar), W/L = 5646, A, 1 eV = 1.6 x 10⁻¹² erg

Ans: 1.9693755 eV. 1.9693755 x (9/10) = 1.7724379543,

[√1.7724379543 = 1.3313 = (4/3), 1.7724379543² = 3.14536 = Pi.]

Same Used ('E' Bar) Cyan = 2.1341802794 eV. Used ('F'Bar) Blue = 2.33121689429 eV. Violet ('G' Bar) = 3.92453623125 eV.

Next: Frequency (or cal.of sec 2) / Calorie (sec.3) = 10 Energy Example : This result is same for all. (we can makes lot of this type Law)

Green = 1.4164 unit (Examp.1)

1.4164 eV = 5.420184477624916e-20 Cal. > (Section 2)

1.4164 J = 0.3383013279831852 Cal. > (Section 3)

5.420184477624916 / 0.3383013279831852 = 16.0217653...It is 10 times of ENERGY.
16.0217653...J = 1.7802 Unit Mass. (M...3)

Orange: 1.1665 unit (Examp.2)

1.1665 eV = 4.463883926256329e-20 Cal. [Frequency & Energy section = 2]

1.1665 Joule = 0.278613738415974 Cal. [Quantum section = 3]

4.463883926256329 / 0.278613738415974 = 16.02176529999999928610053...Unit.

It's 10 times of ENERGY. 16.0217653...J = 1.7802 Unit Mass. (Same result For All)

[When Create one Color From one Mass in that Moment it is Flexible (Quantum Digit, Constant, etc. Frequency, It is depends on Particle's Quality; Mass) For...Exp:4.7515575904251470204270868302612.x.1.2919(Constant)= 6.1385372510702474356897534760144 Frequency of BLUE Light. All most Same of the Past article Chart]

Same Process: Example For ALL Colors : (Same result For all)

Yellow = 1.3331 Unit (Examp.1)

1.3331 Calorie = 5.58142308 Joule. (Section 1, Bar)

1.3331 Joule = 0.3184054647941148 Calorie. (Section 3)

5.58142308 / 0.3184054647941148 = 17.52929424... √17.52929424... = 4.18688 = 4/3 Pi.

(T1 & T2 Group's frequency total 17.44 unit nearer of 17.529...)

Blue = 1.5831 unit (Examp.2)

1.5831 Cal = 6.62812308 J. (Bar.sec.1)

1.5831 J. = 0.3781169389509888 Cal. [Quantum section = 3]

√(6.62812308 / 0.3781169389509888) = √17.529294240... = 4/3 Pi. (4/3 Pi... 2)

SO JOULE (Sec.1) / Calorie (sec.3) = (4/3 Pi)² (Rule)

[## Calorie (sec.2) / Joule (sec.2) = 2.38845, or Frequency / Joule = 2.38845 unit.

Example: For All Colors. Same. Result.

When Red Mass 1 unit.

1 eV = 1.60217653e-19 Joule (section 2)

1 eV = 3.826732898633801e-20 Calorie. (Section 2)

3.826732898633801 / 1.60217653 = 2.38845899662749591020410216594547 Unit.

2.38845⁺³ = 13.62 Bohr 'H' Atom Energy unit.

∛2.38845 = 1.3367 = 4/3, 1.3367⁺² = 1.786755 Unit (M...4)

√2.38845 = 1.5454356 unit Energy, √1.5454356 = 1.2431 (Max. Photon Energy)

CALORIE (sec.2) / eV (sec.1) or Frequency / eV =

Magnetic Field Either Wave Cube = Pi.

Example: For All Colors. Same. Result. (Pi... 1)

Cyan = 1.4997 Unit.

1.4997 Cal = 3.919008824826563e+19 eV. (Section 1)

1.4997 eV = 5.738951328081112e-20 Cal. > This is CYAN Frequency (Sec.2)

5.738951328081112 / 3.919008824826563 = 1.464388467748625481260792608..

1.4643884677486... = 3.1402838 = Pi. 1.4643884677486 x (10/9) = 1.6270983 Unit

eV (Section 3) / Joule (Sec.2) = Red Frequency (R... 1)

Example: For All Colors. Same. Result.

Blue = 1.5831 unit

1.5831 eV = 2.536405664643e-19 J. (Section 2)

1.5831 J. = 9.880933657166979e+18 eV. (Section 3)

9.880933657166979 / 2.536405664643 = 3.895644... Unit (Almost same Red Fqn)

3.895644 / 3 = 1.298548 Unit it is Constant. (Con... 2)

1.298548⁺² = 1.68622695886366... unit is Mass Either energy... (m... 3)

3/1.298548 = 2.31027270459... root = 1.519958125933...; x (10/9) = 1.6888... (m... 4)

2.31027270459 sqrt = 5.33735 (Force ... 2)

3.895644⁺² = 15.176042174736, 15.176042174736 / 9 = 1.686226 (m... 4) E=m·(m/10)

1.298548 x 4 = 5.194192, [see Later it is (root 3)²] (1.29858 x 7 = 9.09)

1.29858/0.413334 (L) = 3.14172073916... = ONE Pi. (Pi... 2)

Cube Root > 5.194192 = 1.731832955424 = Root 3, (√3 ... 1)

1.29858 / (3/4) = Root 3, So Constant / (3/4) = root 3 (√3 ... 2)

[1.29858 / root 3 = 0.75 unit ("r"... 1)]

3.895644... / (1.29858)³ = 1.78... Unit Mass (M... 5 Before create a Nucleus)

SO: This calculation either 3.89.... digit/unit present in all Colors

And Particle.

3.895644... is RED Frequency,

[1 Calorie = 4.1868 Joule, 4.1868 = 4/3 Pi. Or 4.1868 / (4/3) = Pi.]

Red is Basic / Base of Particle to Color either Color to Particle. I have discuss several

time in past ARTICLE. √4.18688 = 1.611746879...⁺² = 2.597728, 2.597728 / 2 = 1.2988,

1.2988, It is Constant. 1.2988⁺² = 1.687 mass. More Prove see later.]

The Quantum Digit are not abstract, 1st example by different method

Image for example (3) 'Atomic Mass and Constant' sree Debasish Dasgupta

4 Calorie x 1.2985 (Constant) = Quantum Digit (Source's)

Either (Root 3) Cube x Calorie, Quantum Digit. [4 x 1.2985... = 5.194... = (root3) Cube]

"Every Quantum Number directs a particular unit or, quantity or position change how much".

This Calculation Done By Each Colors Frequency's JOULE to Equivalent eV.

(As per Table No.1)
 Red fqn. = 3.8757,
 3.8757 Joule equivalent eV = 2.419021829011563 eV.
 2.419021829011563 x 0.413334 = 1 unit,
 Orange fqn. = 4.5216, 4.5216 Joule = 2.822160926299426 eV,
 2.822160926299426 x 0.413334 = 1.1665
 Yellow fqn. = 5.1676, 5.1676 joule = 3.22536243868208 eV
 3.22536243868208 x 0.413334 = 1.3331
 Green fqn. = 5.4905, 5.4905 joule = 3.426900779778618 eV
 3.426900779778618 x 0.413334 = 1.4164
 Cyan fqn. = 5.8135, 5.8135 joule = 3.628501535696947 eV
 3.628501535696947 x 0.413334 = 1.4997
 Blue fqn. = 6.1365, 6.1365 Joule = 3.83102292161276 eV
 3.83102292161276 x 0.413334 = 1.5831]

(Frequency as per Table No. 2)

Red fqn. = 3.8757,
 3.8757 Joule equivalent eV = 2.419021829011563 eV.
 2.419021829011563 Calorie = 6.321376367867027e+19 eV. (A' Group)
 2.419021829011563 Calorie = 10.1279608538932 Joule.
 2.419021829011563 eV = 0.9256950653403168e-19 Cal (B' Group)
 2.419021829011563 eV = 3.875700099566838e-19 Joule (B' Group)
 2.419021829011563 Joule = 0.5777734525547674 Cal. (C' Group)
 2.419021829011563 Joule = 1.509834806503064e+19 eV. (C' Group)

Yellow fqn. = 5.1676,
 5.1676 joule = 3.22536243868208 eV
 3.22536243868208 Calorie = 8.428501607294255e+19 eV (A' Group)
 3.22536243868208 Calorie = 13.50394745827413 Joule.
 3.22536243868208 eV = 1.234260055412246e-19 Cal (B' Group)
 3.22536243868208 eV = 5.16759999999993e-19 Joule.
 3.22536243868208 Joule = 0.7703645836156683 Cal. (C' Group)
 3.22536243868208 Joule = 2.013113023620487e+19 eV. (C' Group)

Same For All Colors. Same calculation; Source's of Quantum Digit

4 Calorie (Group C) x 1.2985 (Con...2) = Quantum Digit

Example: Red : (Frequency as per Table No.2)

2.419021829011563 Joule = 0.5777734525547674 Calorie.
 0.5777734525547674 x 4 x 1.2985 = '3' it is Quantum Digit of Red Frequency.
 (3.895644/3 = 1.298548 Unit it is .Con...2 & Pi x 0.413334 = 1.2985 Con...3
 Already Explain about L = 0.413334 see about it in past article)

Example: Yellow: (Frequency as per Table No.2)

3.22536243868208 Joule = 0.7703645836156683 Calorie
 0.7703645836156683 x 4 x 1.2985 = '4' it is Quantum Digit of Yellow Fqn.
 Like This for All Colors.

Hence: RED: 2.419021829011563 eV = 0.9256950653403168e-19 Cal. (Group : B)
 0.9256950653403168 x 4 x 1.2985 = 4.8 it is Another use Unit For Energy Either
 Charge, So Red Color Energy = 4.8 Unit. [4.8...; ...1]

[Charge on an electron = 4.8 x 10⁻¹⁰ stat C, (as per this process we can
 find out others colours Value unit)]

[Used Visible Light's energy Chart:-

Example: Red 2.419021829011563 Joule = 1.509834806503064e+19 eV. (e...1)

1.509834806503064 x (10/9) = 1.677594229447848..Unit of Mass. As per 'E=m·(m/10),

1.509834806503064⁴ = 5.196581370839, $\sqrt[3]{5.196581370839} = 1.732098467 = \sqrt{3}$.

5.19658.../4 = 1.2991 it is almost same of constant unit 1.2985 ($\sqrt[3]{3} \dots 3$)

[1.509834806503064 / 10 = 0.150983480650364; 0.150983480650364 x 9 = 1.358851325853276 unit Mag
 field or wave, 1.358851325853276 x 7 = 9.511959280972932, $\sqrt[3]{9.511959280972932} = 1.284389...$

1.284389... x 7 = 8.99... or c2, see past article. N.B; AS PER MY CALCULATION Mag field or Wave X
 (10/9) or (M/m)square = Energy, Energy x (10/9) = mass] [1.509...; ...1]

(Why Here use to the power 4 and Cube root,

Because we know that Sphere rule 4/3 Pi x r' cube, So used 4 and 3 ; cube,)

[1 Calorie = 4.1868 Joule, 4.1868 = 4/3 Pi. Or 4.1868 / (4/3) = 3.1401 = ..(Pi...3)

1.509834806503064 x 9 = 13.588 it is Bohr Energy.

$\sqrt[3]{1.509834806503064} = 1.04684178 = 1/3$ Pi. Or 1.04684178 x 3 = Pi. (Pi...4)

Red is Basic / Base of Particle to Color either Color to Particle. I have discuss several

time in past ARTICLE, I think it is enough.]

YELLOW: (Energy either Charge)

[Yellow fqn. = 5.1676, 5.1676 joule = 3.22536243868208 eV.]

3.22536243868208 eV = 1.234260055412246e-19 Cal

1.234260055412246 x 4 x 1.2985 = 6.41 unit (Energy either Charge) for

Yellow Color. Like this we can find others colors Charges.

JOULE (Group B) / CALORIE(B) = 4/3 Pi. Or Frequency / Calorie = 4/3 pi

Example: Same For All Colors. Same result for all colours

RED :

2.419021829011563 eV = 0.9256950653403168e-19 Cal

2.419021829011563 eV = 3.875700099566838e-19 Joule

3.875700099566838 / 0.9256950653403168 = 4.1868 = 4/3 Pi.

OTHERS WAY : Same for all. Same result for all colours

RED:

2.419021829011563 eV = 3.875700099566838e-19 Joule

2.419021829011563 Joule = 0.5777734525547674 Cal

3.875700099566838 (Red Fqn.) / 0.5777734525547674 = 6.708.

$\sqrt[4]{6.708} = 1.60934, 1.60934^3 = 4.16816$ (Nearer 4/3 Pi)

1.60934 (Energy) x (10/9) = 1.788 Unit (M... 6). 1.60934² = 2.5899752356,

2.5899752356 / 2 = 1.295 (Nearer Time x Space 3.6 x 0.36 = 1.296).

1.295² = 1.67699 Unit mass. (m... 5);

$\sqrt[3]{6.708} = 1.8859, 1.8859^4 = 12.649 = 4$ Pi. (Apporx)

12.65 / 4 = 3.1625, 3.1625² = 10.0. $\sqrt{3.1625} = 1.77834$ Unit (M... 7)

[See "Atom to Nucleus" as per this 3.1625 unit Acceleration]

Again : $\frac{3}{4} = 0.75, 0.75^3 = 0.421875, 0.421875 \times 4 = 1.6875$ unit (m... 6)

1.6875 / 3 = 0.5625, 0.5625 x 3.1625 = 1.77890625 and 1.78 Unit (M... 8)

AS PER 4/3 Pi 'r'³ Here r = $\frac{3}{4} = 0.75$, So: 4/3 x $\sqrt[3]{10} \times (3/4)^3 = \text{One}$

Mass. (Before create a nucleus; Please see about digit 10, & others in past article)

0.5625 Cube = 0.177978515625, or 0.178, So 1.78 - 0.178 = 1.602 unit energy.

0.5625 x 2 = 1.125 x 10 = 11.25 it minimum angel of one electron's. see about

past article. 1.125 to the power four = 1.6018 IT is Energy.

[This calculation present in all colours]

Next. Calculation Same for all , same result for all colours

Example : Green fqn. = 5.4905, 5.4905 joule = 3.426900779778618 eV

3.426900779778618 Calorie = 14.34774818477712 Joule

3.426900779778618 Calorie = 8.955160630631081e+19 eV.

3.426900779778618 eV = 1.311383395433267e-19 Calorie.

14.34774818477712 / 1.311383395433267 = 10.94092561 Unit.

$\sqrt{10.94092561} = 3.3$ it is Displacement, (s) (Disp. 1) see E = F/s in past article.

8.955160630631081 / 1.311383395433267 = 6.828789..Unit.

6.828789 x 2 = 13.657578 Unit (Bohr Energy)

6.828789 x 4 = 27.315156, 27.315156 / 3 = 9.105052 = (4/3 x 6.828789) = Unit

(Another Used unit Mass)

$\sqrt{27.315156} = 1.44411$, it is Mag. Filed Either Wave, [1.444...; ...2]

1.44411 x (10/9) = 1.6045671..Unit (E... 2) 1.6045671 x (10/9) = 1.7828524 Unit

Mass, [N.b: *As per This calculation Mag.Field or Wave x 10/9 or (M/m)² =

Energy, Energy x (10/9) = Mass]

[We know Total 50 no. of electron stay in four orbit (32, 8, 8, 2), so 360

Degree for one orbit, 4 x 360 = 1440 Degree total acquire place, as per this

average for one electron 1440 / 50 = 28.8, 28.8 / 2 = 14.4, 14.4 x 9 = 1.6 unit,

Mass either energy for one part, 28.8/4 = 7.2 Unit Binding energy (Bn. 1)]

6.828789 / 2 = 3.4143945, $\sqrt[3]{3.4143945} = 1.50581366$..(Energy) (e...2)

1.50581366 x (10/9) = 1.6731263 Unit. (m... 7)

6.828789 / 3 = 2.276263, $\sqrt{2.276263} = 1.5087289352299$.. Energy (e...3)

1.5087289352299..x (10/9) = 1.676365483588797 Unit Mass. (m... 8)

6.828789 / 6 = 1.1381315, 1.1381315² = 1.29534... (Time x Space)

1.295343311² = 1.6779142941..Unit Mass. (m...9)

1.5087289352299⁴ = 5.18137 unit $\sqrt[3]{5.18137} = \sqrt{3}$. ($\sqrt{3} \dots 4$)

[If we calculate the perimeter of the triangle, we can see every triangle's

perimeter $\sqrt{3} = 1.73205$ unit wise it will increase, see about this in past Article

1.5087289352299 / 10 = 0.15087289352299, 0.15087289352299 x 9 = 1.35785604170691

unit Mag field or wave, 1.35785604170691 x 7 = 9.50499229194837 $\sqrt[3]{9.50499229194837} =$

1.28428... 1.28428 x 7 = 8.9899 or c2; N.B; AS PER MY CALCULATION Mag field or Wave X

(10/9) or (M/m)square = Energy, Energy x (10/9) = mass $\sqrt[3]{1.5087289352299} = 1.046756559$

1.04675656 = 1/3 Pi; or 1.04675656 x 3 = Pi] [1.509...; ...2]

Electron Volt / Joule = 6241509479607718382.942483872236 unit.

81root 6241509479607718382.942483872236 = 1.70624193966795373888...;

1.7062419396679...; x 2 = 3.412483879...; Cube rt 3.412483879 = 1.505532... (E)

1.505532...x (10/9) = 1.6728141497578193322192793776545 (m10)

- Inside of Each and Every particle and energy have 3 parts or sections; There are present value of 4/3; 3/4; Pi; Root 3; 10; 9; 3.89 etc. and must have anothers pasonal Bars (Constant like Honorable Scientist PLANK).
- Mag.Field or Wave x 10/9 or (M/m)² = Energy, Energy x (10/9) = Mass] ■ 1.298548 Unit it is Constant. And Pi x 0.413334 = 1.2985

